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~~8.1 Describing Chemical Reactions (1/2)~~ **How to Write Balanced Chemical Equations From Words - TUTOR HOTLINE**

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Chemical Reactions 8.1 Describing
 Chemical Reactions (2/2) Introduction to
 Balancing Chemical Equations
 Describing Chemical Reactions
 Describing Chemical Reactions Types of
 Chemical Reactions **Describing
 Chemical Reactions** Describing
 Chemical Reactions **How to Write
 Chemical Equations** Writing and
 Balancing Reactions Predicting Products
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 Workbook CHAPTER 11, Chemical
 Reactions(continued) 9 Use the symbols

in Table 11.1 on page 323 to write a skeleton equation for the following chemical reaction Hydrochloric acid reacts with zinc to produce aqueous

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The new substances formed as a result of a chemical reaction are called products. 2. Circle the letter of each sentence that is a correct interpretation of the chemical equation $C + O_2 \rightarrow CO_2$.
a. Carbon and oxygen react and form carbon monoxide. b. Carbon and oxygen react and form carbon dioxide. c. Carbon dioxide yields carbon and oxygen. d. The reaction of carbon and oxygen yields carbon dioxide.

Chapter 7 Chemical Reactions

Section 7.1 Describing Reactions

Chemical reactions take place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread. Chemists use a chemical equation—a quick, shorthand notation—to convey as much information as possible about what happens in a chemical reaction. Figure 11.1 Chemical changes occur when bread

11.1 Describing Chemical Reactions

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A Chemical Reaction Occurs When One Or More Chapter 11 ...

A double-replacement reaction is a reaction in which the positive and negative ions of two ionic compounds exchange places to form two new compounds. Contributors and Attributions CK-12 Foundation by Sharon Bewick, Richard Parsons, Therese Forsythe, Shonna Robinson, and Jean Dupon.

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Chapter 4 Atomic Structure Chapter 8
Photosynthesis, TE SECTION 11.1
DESCRIBING CHEMICAL REACTIONS
(pages 321–329)

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Chapter 11 Chemical Reactions Workbook Answers

In a chemical reaction, one or more
substances (the reactants) change into

one or more new substances (the
products). Figure 11.1a shows the
ingredients for making leavened
bread—flour, salt, yeast, and water.
These are the reactants in a chemical
reaction that takes place when the
ingredients are mixed together and
heated in the oven.

11.1 Describing Chemical Reactions 11

A useful description of a chemical
reaction tells you the substances present
before and after the reaction. In a
chemical reaction, the substances that
undergo change are called reactants.
The new substances formed as a result
of that change are called products. In
Figure 1, the reactants are the carbon in
the charcoal and the oxygen in the

air. The product of the reaction is carbon dioxide gas.

Section 7.1 7.1 Describing Reactions

What Is a Chemical Reaction? A chemical reaction is a process in which some substances change into different

substances. Substances that start a chemical reaction are called reactants. Substances that are produced in the reaction are called products. Reactants and products can be elements or compounds.