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 density of lead is 11.4 g/cm³. The mass
 of a lead ball with a radius of 0.50 mm is
 _____ g.($\pi=3.1416$; Volume sphere
 $= \frac{4\pi r^3}{3}$) AP CHEMISTRY CHAPTER 1

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 the elements: lead, silver, gold,
 potassium, mercury, iron, and sodium.
 What are the SI units for: mass, length,
 volume, energy, time? The density of
 aluminum is 2.7 g/cm³. A piece of
 aluminum that occupies a volume of

21.4 cm³ ...

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The first ionization energy for a neutral atom of chlorine is $1.25 \text{ MJ} / \text{mol}$ and the first ionization energy for a neutral atom of argon is $1.52 \text{ MJ} / \text{mol}$. How would the first ionization energy value for a neutral atom of potassium compare to those values?

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% b. 6.6 % c. 3.3 % d. 5.0×10^{-4} % e. 2.2 % 2. A quantitative observation a. contains a number and a unit. b. does not contain a number.

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Baker, Mrs. (Science) / AP Chemistry

1.1 Chemistry: An Overview A. Reaction of hydrogen and oxygen 1. Two molecules of hydrogen react with one molecule of oxygen to form two molecules of water $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 2. Decomposition of water $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$ B. Problem Solving in Chemistry (and life) 1. Making observations 2. Making a prediction 3.

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test version a Name: Date: 1. A. How many moles of oxygen are in 3.24 grams of Calcium Nitrate. B. How many grams of oxygen are in the same sample? C. What is the mass of 7.32×10^{24} molecules of KCl? 2. A. Balance the following reaction $_\text{HNO}_3 + _\text{Ba}(\text{OH})_2 \rightarrow _\text{Ba}(\text{NO}_3)_2 + _\text{H}_2\text{O}$ B.

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Chapter 2: Atoms, Molecules, Ions.
Formulas sheet for tests. Quizzes:
Monatomic Ions Polyatomic Ions Prefixes & Common Acids Measurement & Dimensional Analysis Measurement & Dimensional Analysis (answers) Practice Test: Practice test for Chs. 1 & 2. Problem set, Ch. 1. #27-87 odd, but skip 39-55. Problem set, Ch ...

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