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The Behavior of Gases 419 Sample Problem 14.1 Answers 7. 105 kPa $\times$ 2.50 L = 40.5 kPa $\times$ V2 V2 = 105 kPa $\times$ 2.50 L/40.5 kPa = 6.48 L 8. 205 kPa $\times$ 4.00 L = P2 $\times$ 12.0 L P2 = 205 kPa $\times$ 4.00 L/12.0 L = 68.3 kPa Practice Problems Plus The volume of a gas at 99.6 kPa and 24°C is 4.23 L.

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Piersa, Amanda / Behavior of Gases

1.Gas particles are so small in relation to the distances between them that their individual volumes can be assumed to be insignificant.2. No attractive or repulsive forces exist between gas particles.3.Gas particles are in constant random motion, traveling in independent straight paths until they collide with each other or other objects.

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