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10. Using only your answers to questions 7-9, calculate how many Joules of energy it takes. to change 12.7 g of water at 22oC to steam at 159oC. Questions 7-9 are each different "steps" for taking 12.7g of water from 22oC to 159oC so. you can simply add the answers: $4140.7 + 28,702 + 1513.6 = 34,356.3$ J. 11.

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Let $\vec{V}_1 = -6.3 \hat{i} + 8.0 \hat{j}$ and $\vec{V}_2 = -4.9 \hat{j} + 4.3 \hat{i}$. A) Determine the magnitude of $\vec{V}_1$. B) Determine the direction of $\vec{V}_1$. C) Determine the magnitude of $\vec{V}_2$. C) Determine the direction of $\vec{V}_2$.