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Key Questions 10.
Using your general
scheme for calculating
the concentration of
diluted solutions
devised in Q1,
calculate the resulting
concentration (ppm
Zn) for each solution A-
E above. Enter your
answers into the
diagram in Figure 1
and into the boxes in
Table 1. Verify answers
provided in Table 1.
Figure 1 Preparation of
calibration

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Interpreting Solubility Curves Model Table1. Solubility Data Temperature (°C)
Solute (g) per 100g of H₂O (l) 10 33 30 42 50 52 70 62 90 73 Task
Complete the model:
Using the grid below make a graph of the solubility data in Table 1. a. Label x-axis, y-axis and create appropriate scales for each. b. Plot points using a pencil Key Questions 1.

Interpreting Solubility Curves

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 altered from the
 properties of their
 individual components.
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Key Question 1. What
 is the mass of carbon
 in 16.0 g of methane?
 12.0 g C 2. Show the
 set-up that would be
 used to determine the
 percent by mass of

carbon in methane in
 order to arrive at the
 answer shown in the
 Model 3 table. 100
 $16.0 \text{ g} \times \frac{12.0 \text{ g C}}{16.0 \text{ g CH}_4} = 12.0 \text{ g C}$
 What information do
 you need in order to
 determine the percent
 composition by

Instructors Guide:
Percent Composition

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 of the solutions for you
 to be successful. As
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 suggest that you have
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