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Chapter 22 Hydrocarbons

You are at chapter 22!! Let's goooo!!!!
 First of all, a hydrocarbon is the simplest organic compound and it only contains carbon and hydrogen. For this, we can use our background knowledge to know that since carbon has four valence electrons, it can form four covalent bonds. Also, since hydrogen only has one, it can only make one bond.

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atom. Therefore, the simplest hydrocarbon molecule consists of a carbon atom bonded to four hydrogen atoms, CH₄. This substance is called methane and is an excellent fuel and the main component of natural gas. 698 Chapter 22 Hydrocarbons Figure 22-1 Carbon is a nonmetal element located in group 4A of the periodic table. Elements in this

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a controlled process by which hydrocarbons are broken down or rearranged into smaller, more useful molecules. Related Flashcards Chapter 22 Organic Chemistry Vocabulary

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786 Chapter 22 • Substituted Hydrocarbons and Their Reactions Section 22.12.1 Figure 22.1 All of these items contain at least one of the functional groups that you will study in this chapter. For example, the fruit and flowers have sweet-smelling aromas that are due to ester molecules. Objectives Define functional group, and give examples.

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This unit looks at aliphatic and aromatic hydrocarbons, including alkyl substituted hydrocarbons and structural and geometric isomers. There is also a brief introduction to functional groups. Chapter 22 - Hydrocarbon Compounds

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Chapter 21: Hydrocarbons

Chapter 22 Hydrocarbons 700 Chapter 22 Hydrocarbons Figure 22-4 This taxi in Bangkok, Thailand is one of many cars, trucks, and buses that have been modified to burn propane. Burning propane benefits the environ-ment because it produces less air pollution than does gasoline. In Table

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