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Introduction to Acids and Bases pages 634-643 Practice Problems pages 635-640 1. Write balanced equations for reactions between the following. a.
aluminum and sulfuric acid $2\text{Al(s)} + 3\text{H}_2\text{SO}_4\text{(aq)} \rightarrow \text{Al}_2\text{(SO}_4)_3\text{(s)} + 3\text{H}_2\text{(g)}$

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