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 17,1977 the fact that
 nitrogen at -195°C is
 about 5% imperfect at
 atmospheric pressure,
 the correction being a
 linear function of the
 pressure. Instead of a
 curve with kinks a
 smooth S-shaped
 isotherm was obtained

with nitrogen at —
195°C.

**Brunauer S, Emmett
P H & Teller E.
Adsorption of gases
in ...**

Paul Hugh Emmett was an American chemist best known for his pioneering work in the field of catalysis and for his work on the Manhattan Project during World War II. He spearheaded the research to separate isotopes of uranium and to develop a corrosive uranium gas. Emmett also made significant contributions to BET Theory which explains the relationship between surface area and gas adsorption. He served on the faculty of Johns Hopkins University for 23 years throughout his scientific career.

**Paul Hugh Emmett -
Wikipedia**

S. Brunauer, P. H. Emmett and E. Teller, "Adsorption of Gases in Multimolecular Layers," Journal of the American Chemical Society, Vol. 60, No. 2, 1938, pp. 309-319.

**S. Brunauer, P. H.
Emmett and E.
Teller, "Adsorption
of ...**

The gas adsorption technique is based on the Brunauer-Emmett-Teller (BET) model included in the determination of the amount of gas that is adsorbed on the surface of the sample at low temperatures [13]. The specific surface area (m^2/g) defined by this method comprises the outward as well as the internal (open pores) surface area. However, the

surface area of closed pores cannot be determined because of its inaccessibility for adsorbing gas molecules.

Brunauer-Emmett-Teller Model - an overview |

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brunauer s emmett p h teller e adsorption of gases in is available in our digital library an online access to it is set as public so you can get it instantly. Our digital library hosts in multiple locations, allowing you to get the most less latency time to download any of our books like this one. Kindly say, the brunauer s emmett p h teller e adsorption of gases in is

Brunauer S Emmett P H Teller E Adsorption Of Gases

In

By making use of the evaporation-condensation properties of liquid mixtures, the Brunauer-Emmett-Teller theory of multimolecular adsorption is extended to mixtures of gases. No satisfactory experimental data are available at the present time with which to test the theory.

Theory of Multimolecular Adsorption from a Mixture of ...

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Accumulation of alkali
promoters on surfaces
of iron synthetic
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Journal of the American
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310-315. 1: 1937

Paul Hugh Emmett - Publications

310 STEPHEN
BRUNAUER, P. H.
EMMETT AND EDWARD
TELLER Vol. 60
DeB~er~ in several
papers showed that
various experimental
adsorption isotherms
could be fitted by
equation (4a).
However, they could
not evaluate K₁
because in all cases,
except one, the
surface, and therefore
v_m, was not known.

[CONTRIBUTION FROM THE BUREAU OF Adsorption Gases in ...

P. H. Emmett's 15
research works with
1,206 citations and 356
reads, including: THE
APPLICATION OF
POLANYI'S POTENTIAL
THEORY TO THE VAN
DER WAALS
ADSORPTION OF
GASES ON IRON
SYNTHETIC AMMONIA
...

P. H. Emmett's research works

Brunauer-Emmett-Tell
er theory aims to
explain the physical
adsorption of gas
molecules on a solid
surface and serves as
the basis for an
important analysis
technique for the
measurement of the
specific surface area of
materials. The
observations are very
often referred to as
physical adsorption or
physisorption. In 1938,
Stephen Brunauer, Paul

Hugh Emmett, and Edward Teller published the first article about the BET theory in the Journal of the American Chemical Society. The BET theory ...

BET theory - Wikipedia

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Matthew S. Cummings,
Ryan Luebke, Martyn
S. Brown, Silvia Favero,
Martin P. Atfield, Flor
Siperstein, David
Fairen-Jimenez, Klaus
Hellgardt, Russell
Purves, David Law,
Camille Petit.
Screening
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Complementary
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Measurement
Techniques.

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Emmett, P.H. and
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Access Free Brunauer S
Emmett P H Teller E
Adsorption Of Gases
Incoun), by latest
(which means date of
upload), or by random
(which is a great way
to find new material to
read). Brunauer S
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S, Emmett P H & Teller
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E., "Adsorption of ...**
The
Brunauer-Emmet-Telle
r (BET) equation
(Brunauer et al., 1938)

overcomes the
limitation to a
monolayer of sorbate
and currently is a
commonly used model
to describe the
multilayer adsorption
of gas molecules on a
sorbent (Brunauer et
al., 1938). For water
adsorption, the BET
equation can be
written as