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$(F-Cl) = (2.30-1.59) = 0.71$ and $(O-S) = 0.54$. So, $(N-P) \sim 0.37$. The AVEE of N is 1.82, so we predict an AVEE for P of 1.45. The actual value is fairly close, 1.42. Chem;Gl;Answers ChemActivity 4 1. No. The ionization energy of He would be about $4\times$ (twice the charge and half the distance) the ...

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