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Balancing Chemical Equations Worksheet 1

Balancing Chemical Equations

Worksheet 1 - ANSWERS 1. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 2. $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$ 3. $\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$ 4. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ 5. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 6. $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$ 7. $2\text{Li} + \text{F}_2 \rightarrow 2\text{LiF}$ 8. $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$ 9. $2\text{NH}_3 \rightarrow \text{N}_2 + 3\text{H}_2$ 10. $2\text{HCl} \rightarrow \text{H}_2 + \text{Cl}_2$ 11. $2\text{NI}_3 \rightarrow \text{N}_2 + 3\text{I}_2$ 12. $2\text{HI} \rightarrow \text{H}_2 + \text{I}_2$ 13. $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$ 14.

Balancing Chemical Equations Worksheet 1

Balancing_Chemical_Equations_Workshe
et._ - 1 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ 2 $2\text{NaF} + \text{Br}_2 \rightarrow 2\text{NaBr} + \text{F}_2$ 3 $2\text{KCl} + 2\text{KCl} \rightarrow 4\text{KCl}$ 4 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$ 5 $\text{P}_4 + 3\text{O}_2 \rightarrow 2\text{P}_2\text{O}_3$ 6 $\text{Mg} + \text{H}_2\text{SO}_4$

Balancing_Chemical_Equations_Work sheet._ - 1 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$...

Balancing Chemical Equations Answer
Key Balance the equations below: 1) $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ 3) $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$ 4) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 5) $\text{Pb}(\text{OH})_2 + 2\text{HCl} \rightarrow \text{PbCl}_2 + 2\text{H}_2\text{O}$ 6) $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2(\text{SO}_4)_3$ 7) $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$ 8) $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$ 9) $2\text{C}_8\text{H}_{18} + 25\text{O}_2 \rightarrow 16\text{CO}_2 + 18\text{H}_2\text{O}$ 10) $\text{FeCl}_3 + 3\text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + 3\text{NaCl}$

NaCl

Chapter 7 Worksheet #1 Balancing Chemical Equations

Title: Balancing Equations Worksheet #1
- ANSWERS Author: James Last modified
by: James Created Date: 11/13/2010
5:46:00 PM Other titles: Balancing
Equations Worksheet #1 - ANSWERS

Balancing Equations Worksheet #1 - ANSWERS

If you also get perplexed in balancing
chemical equations, follow the tips for
correct balancing chemical equations
worksheet answers. Tip # 1: When you
are trying to balance the chemical
equations, you should remember that
you can only change the value of
coefficient in front of the element or
compound, and not the subscript.

49 Balancing Chemical Equations Worksheets [with Answers]

1. Balanced equations. (Coefficients
equal to one (1) do not need to be
shown in your answers). (a)
 $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$. (b)
 $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$. (c)
 $2\text{FeBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 6\text{HBr}$
(d) $\text{C}_4\text{H}_{10}\text{O}_3 + \text{H}_2\text{O} \rightarrow 2\text{C}_2\text{H}_4\text{O}_2$. (e)
 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ (f)
 $\text{C}_4\text{H}_{10}\text{O} + 6\text{O}_2 \rightarrow 4\text{CO}_2 + 5\text{H}_2\text{O}$ (g)
 $\text{C}_7\text{H}_{16} + 11\text{O}_2 \rightarrow 7\text{CO}_2 + 8\text{H}_2\text{O}$ (h)
 $4\text{H}_2\text{SiCl}_2 + 4\text{H}_2\text{O} \rightarrow \text{H}_8\text{Si}_4\text{O}_4 + 8\text{HCl}$
(i) $10\text{HSiCl}_3 + 15\text{H}_2\text{O} \rightarrow 10\text{H}_2\text{SiO}_3 + 30\text{HCl}$ (j)
 $\text{C}_7\text{H}_9 + 3\text{HNO}_3 \rightarrow \text{C}_7\text{H}_6(\text{NO}_2)_3 + 3\text{H}_2\text{O}$ (k) $\text{C}_5\text{H}_8\text{O}_2 + 2\text{NaH} + 2\text{HCl} \rightarrow \dots$

Balancing Equations: Practice Problems

W 301 Everett Community College
Tutoring Center Balancing Equations
Worksheet - Solutions . 1) $\text{H}_3\text{PO}_4 + 3\text{KOH} \rightarrow \text{K}_3\text{PO}_4 + 3\text{H}_2\text{O}$.

Balancing Chemical Equations Worksheet 1 Answer Key

Balancing Chemical Equations
Worksheet + O Æ H O Pages 1 - 2
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Balancing Chemical Equations Worksheet | Free Resume Templates

Name ____ Class ____ Date ____
Balancing Chemical Equations
Worksheet For the following: 1. Draw a
circle around each subscript. 2. Draw a
square around each coefficient. H_2O
 $5Cl_2$ $2Mg$ $3H_2O_2$ For the following 1.
List the chemical symbols of each
element. 2. Give the number of atoms of

each element.

Balancing Chemical Equations Worksheet_8 (2).doc - Name Class ...

This worksheet includes some rules and
guidelines to help you balance chemical
equations. Rules 1.) The formulas of the
reactants and products cannot be
changed, do not alter subscripts or
charges. 2.) The only numbers that can
be changed are the numbers indicating
how many molecules or atoms, which
are called coefficients. 3.)

Balancing Chemical Reactions

Balancing Equations Worksheet and Key
1. Answer the following questions about
the chemical equation shown below: $2H_2 + O_2 \rightarrow 2H_2O$... Balance the
following chemical equations: a) $Fe + O_2 \rightarrow Fe_2O_3$ b) $H_2 + Cl_2 \rightarrow HCl$ c) $Ag + H_2S \rightarrow Ag_2S + H_2$ d) $CH_4 + O_2 \rightarrow CO_2 + H_2O$ e) $HgO \rightarrow Hg + O_2$...