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This process is summarized by a chemical equation. In the Balancing

Chemical Equations Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$. 1. How many atoms are in a hydrogen molecule (H_2)? 2. How many atoms are in an oxygen molecule (O_2)? 3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)?

Balancing_Chemical_Equations_Student_Exploration_Gizmolab ...

The two chemical equations each have an arrow in them. The arrow represents a chemical change. Use the pieces and equations to try and balance each of the formulas so that there is the same number of atoms on each side of the chemical reaction. Use the labels to label the reactants and products on your 2 equations.

2_Exploration_- _Single_Student_Online_- _Balancing_Chemical_...

Student Exploration: Balancing Chemical Equations. DEFINE Vocabulary: coefficient. Combination- a chemical reaction when two reactions combine to make one. Compound- a substance consisting of two or more elements . Decomposition- the separation of a substance into a simpler form . Double replacement- multiple elements that are getting swapped

BalancingChemEquationsSE.docx

The equation $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ is unbalanced because there are two oxygen atoms on the reactants side of the equation, and only one on the products. side of the equation. To

balance the equation, you cannot change the structure of any of the molecules, but you can change the number of molecules that are used.

Student Exploration Sheet: Growing Plants

This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \Rightarrow \text{H}_2\text{O}$. How many atoms are in a hydrogen molecule (H_2)?

Balancing Equations Gizmo - Student Exploration Balancing ...

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As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Student Exploration: Balancing Chemical Equations In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$. How many atoms are in a hydrogen molecule (H_2)?

Student Exploration Balancing Chemical Equations Gizmo ...

Balancing Chemical Equations. Balance and classify five types of chemical reactions: synthesis, decomposition, single replacement, double replacement, and combustion. While balancing the reactions, the number of atoms on each side is presented as visual, histogram, and numerical data. Use for 5 minutes a

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The first step to balance the equation is
to write down the chemical formula of

reactants that are listed on Page 23/30
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Balancing Chemical Equationsthe left
side of the chemical equation.

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Chemical Equations**

To balance a chemical equation, you first
need to be able to count how many
atoms of each element are on each side
of the equation. In this activity, you will
practice counting the atoms that are
represented in chemical formulas.