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proton: Subatomic particle with one unit of positive charge. electron: Subatomic particle with one unit of negative charge. atomic number: Indicates the number of protons in the nucleus of a specific element, as well as the number of electrons in an electrically neutral atom. atomic mass: An approximation of the total mass of an atom, measured in Daltons.

Chapter 2: The Chemical Context of Life

Name: Roksana Korbi____AP Biology Chapter 2 Active Reading Guide The Chemical Context of Life This chapter covers the basics that you may have learned in your chemistry class. Whether your teacher goes over this chapter, or assigns it for you to review on your own, the questions that follow should help you focus on the most important points. ...

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AP Biology Reading Guide Fred and Theresa Holtzclaw Chapter14: Mendel and the Gene Idea 12. 13. 14. As you start to work word problems in genetics, two things are critical: the parent's genotype must be correct. and the gametes must be formed correctly. Using Figure 14.8 as your guide, explain how the gametes are derived for the following cross.

Leology - Welcome

AP Biology Reading Guide Fred and Theresa Holtzclaw Chapter 16: Molecular Basis Of Inheritance 20. 21. 22. 23. Explain the rule. to a d,ame+cr. Describe the structure of DNA relative to each of the following: a. distance across molecule b. distance between nucleotides - H c. distance between turns d. components of the backtx-)ne

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Victoria Brown | AP Biology

Using Figure 21.7 as a guide, label the types of DNA sequences in the human genome and give their percentages. AP Biology Reading Guide Chapter 21: Genomes and Their Evolution