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Chapter 11: Chemical Reactions

Chapter 11: Chemical Reactions Study Guide □chemical equation A representation of a chemical reaction with reactants on the left, products on the right, and an arrow separating the two skeleton

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CK-12 Chemistry - Intermediate

Section 11.1 –Describing Chemical Reactions. The reactants are written on the left and the products on the right. The arrow that separates them is called yield. Reactants Products. Symbols in Equations. Symbol Meaning " yields D reversible reaction (s) solid (l) liquid (g) gas (aq) aqueous " catalyst " heat. Pt .

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To describe a chemical reaction. To calculate the quantities of compounds produced or consumed in a chemical reaction. As shown in Figure 11.3.1, applying a small amount of heat to a pile of orange ammonium dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano.

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balanced equation. A _____ is an equation where each side of the equation has the same number of atoms of each element, and mass is conserved. word equation. Iron + oxygen ---> iron (III) oxide is an example of a(n) _____. chemical equation. Fe + O2 ---> Fe2O3 is an example of a(n) _____.

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