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~~Coffee Cup Calorimetry~~

3 L1 Specific heat

calorimetry lab

Gizmo Calorimetry Lab Answers

Gizmo Warm-up A
calorimeter is an
insulated container
filled with a liquid,
usually water. When a

hot object is placed in
the calorimeter, heat
energy is transferred
from the object to the
water and the water
heats up. Calorimeters
can be used to find a
substance's specific
heat capacity.

Student Exploration- Calorimetry Lab (ANSWER KEY)

1 Calorimetry Lab

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temperature receptor,
3 is a light touch
receptor, and 4 is a
strong pressure
receptor [Book]
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Quiz Answers
Investigate how
calorimetry can be
used to find relative
specific heat values
when different
substances are mixed
with water.

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Quiz Answers - HPD
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Student Exploration-
Calorimetry Lab
(ANSWER KEY) June 04,
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Student Exploration:
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Vocabulary: calorie,
calorimeter, joule,
specific heat capacity
Prior Knowledge
Questions (Do these
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calorimeter and heated
2,000 g of water by 20
° C. How much thermal
energy has been
released? (Hint: Use a
specific heat equation.)
Give your answer in
both joules and
calories. C.How many
kilocalories does the
snack bar contain? In
this Gizmo, you will

investigate how calorimetry can be used to find relatively

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

Student Exploration: Determining a Spring Constant (ANSWER ...

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When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper.

CalorimetryLabSE.1.pdf - Name Date Student Exploration ...

Student Exploration: Calorimetry Lab Vocabulary: calorie,

calorimeter, joule,
specific heat capacity

Prior Knowledge

Questions (Do these

BEFORE using the

Gizmo.) 1. The Latin

word calor means

“heat,” and meter

comes from the Greek

word meaning “to

measure.” What do you

think a calorimeter

does? Measures heat 2.

Where have you heard

the word calorie

before? What do you

think a calorie is?

Calorimetry Lab SE -

Student Exploration

Calorimetry Lab ...

Correct Answer: C. The

final temperature of

the lead-water system

will be lower than the

final temperature of

the copper-water

system. A blacksmith

heats a 1,540 g iron

horseshoe to a

temperature of 1445°C

before dropping it into

4,280 g of water at
23.1°C.

Calorimetry Lab

Flashcards | Quizlet

When a hot object is

placed in the

calorimeter, heat

energy is transferred

from the object to the

water and the water

heats up. Calorimeters

can be used to find a

substance’s specific

heat capacity. You will

use the Calorimetry

Lab Gizmo to

determine the specific

heat capacities of

various substances. 1.

On the SIMULATION

pane, select Copper.

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cx - Name_Ehwnett

Jonas_Date Student

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Calorimetry Lab Gizmo

Answers The

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Calorimetry Lab

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be

mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

Calorimetry Lab

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

Calorimetry Lab

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Calorimetry Lab Gizmo

Answer Key

Calorimetry Lab

Calorimeters can be

used to find a

substance's specific

heat capacity. You will

use the Calorimetry

Lab Gizmo™ to

determine the specific

heat capacities of

various substances. 1.

On the SIMULATION

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Student Exploration-

Calorimetry Lab

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2 Help Calorimetry Lab

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Calorimetry Lab

Calorimeters can be

used to find a

substance's specific

heat capacity. You will

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Gizmo Warm-up A

calorimeter is an

insulated container

filled with a liquid,

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from the object to the

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can be used to find a

substance's specific

heat capacity.

Student Exploration:

Calorimetry Lab

This project was created with Explain Everything™ Interactive Whiteboard for iPad.

**Calorimetry Gizmo
Part 2 Help -
YouTube**

You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances.

1. On the SIMULATION pane, select Copper. Use the slider to set its Mass to 200 g.