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 volume b. molar mass c. atomic mass
 ____ 1. the number of grams of an
 element that is numerically equal to the
 atomic mass of the element in amu ____
 2. the mass of a mole of any element or
 compound ____ 3. the volume occupied
 by a mole of any gas at STP

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 Section 10.1 Measuring Matter 1. pair 2.
 5 3. dozen 4. gross 5. 200 6. ream 7.
 6,000,000,000 8. 0.5 mol 9. 6.02 10²³
 10. four moles 11. 6.02 10²³ Cu atoms
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 LLabab How much is a mole? Counting
 large numbers of items is easier when
 you use counting units such as decades
 or dozens. Chemists use a counting unit
 called the mole. Procedure 1. Read and
 complete the lab safety form. 2. Select
 an item to measure, such as a paper
 clip, gum

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CHAPTER 10 - STUDY GUIDE Summary of
 the Chapter All landforms are composed
 of rocks or their weathered by products.
 Three main types of rocks can be
 identified on the Earth's surface:
 igneous, sedimentary and metamorphic.
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 Balance is influenced by: age, inactivity,
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 situation of the United States in 1791.

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SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287-296) This section defines the mole and explains how the mole is used to measure matter. It also teaches you how to calculate the mass of a mole of any substance. Measuring Matter (pages 287-289) 1.

SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287-296)

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Summary Chapter 11 - "Like Summer Tempests Came His Tears" Now that Toad is back at the river, he wants to return immediately to Toad Hall. However, Rat has bad news: weasels and stoats (creatures from the Wild Wood) moved into the property when they learned about Toad's sentence, and are now squatting there. Badger and Mole had tried to save the house, but were chased away by the weasels and ...