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Section 1 Properties of Matter. 1. mass. 2. substance. 3. properties. 4. Physical. 5. density. 6. chemical. 7. physical. 8. chemical. 9. chemical. 10. physical. 11. physical. 12. chemical. 13. physical. 14. physical. 15. physical. 16. physical. 17. physical. 18. gas. 19. solid. 20. liquid. 21. true. 22. false. 23. true. 24. false. 25. false. 26. false. 27. false. 28. false Section 2 Changes in Matter . 1. boil. 2. freeze. 3. condense. 4.

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Because the physical universe is so vast, there are many different branches of science (Figure 1.1). Thus, chemistry is the study of matter, biology is the study of living things, and geology is the study of rocks and the earth. Mathematics is the language of science, and we will use it to communicate some of the ideas of chemistry.

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There are many possible definitions of matter. In science, matter is the term for any type of material. Matter is anything that has mass and takes up space. At a minimum, matter requires at least one subatomic particle, although most matter consists of atoms. The word matter is sometimes used to refer to a pure substance .

What Is Matter? Definition in Chemistry

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