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in the space provided. 1. a. Write the

two equations that show the two-stage

ionization of sulfurous acid in water.

stage 1: $\text{H}_2\text{SO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq})$

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