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Chemistry 8 Assessment Ionic Compounds

210 Chapter 8 Ionic Compounds CHAPTER 8 What You'll Learn You will define a chemical bond. You will describe how ions form. You will identify ionic bonding and the character-istics of ionic compounds. You will name and write formulas for ionic compounds. You will relate metallic bonds to the characteristics of metals. Why It's Important

Chapter 8: Ionic Compounds

in the chemical formula for any ionic compound the chemical symbol for the ____ is written first followed by the chemical symbol of the _____. cation, anion. a small number used to represent the number of ions of a given element in a chemical formula and are written to the lower right of the symbol for the element.

Chemistry Chapter 8 Ionic Bonds Flashcards | Quizlet

Ionic Compounds Section 8.1 Forming Chemical Bonds In your textbook, read about chemical bonds and formation of ions. Use each of the terms below just once to complete the passage. energy-level The force that holds two atoms together is called a(n) (1) Such an attachment may form by the attraction of the positively charged (3) eleay-vns

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A chemical bond consisting of a hydrogen atom between two electronegative atoms (e.g., oxygen or nitrogen) with one side being a covalent bond and the other being an ionic bond; the attractive force between the hydrogen attached to an electronegative atom of one molecule and an electronegative atom of a different molecule.

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combine to form an ionic compound. Group 1 Group 15 Three group 1 atoms loses 1 e, forming 1 ions. One group 15 atom gains 3 e, forming a 3 ion. The ions attract, forming X₃Y, where X represents a group 1 atom and Y represents a group 15 atom. Section 7.2 Assessment page 217 12. Explain how an ionic compound made up of

Ionic Compounds and MetalsIonic Compounds and Metals

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Recognizing Compounds With Ionic Bonds . You can recognize ionic compounds because they consist of a metal bonded to a nonmetal. Ionic bonds form between two atoms that have different electronegativity values.Because the ability to attract electrons is so different between the atoms, it's

like one atom donates its electron to the other atom in the chemical bond.

Examples of Ionic Bonds and Compounds - ThoughtCo

Ionic compounds contain both cations and anions in a ratio that results in no net electrical charge. In covalent compounds, electrons are shared between bonded atoms and are simultaneously attracted to more than one nucleus. In contrast, ionic compounds contain cations and anions rather than discrete neutral molecules.

2.7: Ions and Ionic Compounds - Chemistry LibreTexts

Melting an ionic compound also frees the ions to conduct a current. Ionic compounds conduct an electric current when melted or dissolved in water. Figure $\text{\PageIndex{4}}$: In an ionic solution, the $\text{\ce{A^{+}}}$ ions migrate toward the negative electrode, while the $\text{\ce{B^{-}}}$ ions migrate toward the positive electrode.

8.9: Physical Properties of Ionic Compounds - Chemistry ...

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2. Chemical bonding. 2.1 Chemical Compounds. Depth of treatment. Compounds. Simple chemical formulas. 2.2 Ionic Bonding. Activities. Representation of ionic bonds using dot and cross diagrams. Examination of a model of the NaCl crystal. Depth of treatment. Positive and negative ions. Sodium chloride crystal structure.

How do atoms form the ions in ionic compounds? | 16-18 ...

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