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f. N in HNO_3 4 g. C in $(\text{HCO}_3)_3$ 3 h. N in $(\text{NH}_4)_2\text{SO}_4$ 2 a.

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Chapter 7 Chemical Formulas And Compounds Test B Answers

WRITING CHEMICAL FORMULAS PRACTICE • Write the formula of the binary ionic compound that will result between the following elements: • 1. zinc and sulfur • 2. aluminum and phosphorus • 3. strontium and chlorine • 4. barium and oxygen • 5. sodium and nitrogen • 6. lithium and iodine

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with which they undergo certain chemical reactions. 20. A(n) uses symbols and formulas to represent the names and relative amounts of reactants and products in a chemical reaction. 21. The law of must be satisfied for a chemical equation to be balanced. 22. A(n) product is sometimes indicated by an arrow

pointing upward. 23.

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