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The gas particles take up more volume relative to the overall volume. The gas particles become hotter, so they increase the pressure along the container wall.

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Displaying top 8 worksheets found for - Gas Behavior. Some of the worksheets for this concept are Unit 2 work 4 gas behavior exercises, Behavior of gases workbook answers, Vocabulary, Section behavior of gases answer key, Behavior of liquids and solids lab answers, 9 05ab episode 901 review wkst key, Behavior of gases guided practice problems answers, Properties of gases.

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Piersa, Amanda / Behavior of Gases

1. Gas particles are so small in relation to the distances between them that their individual volumes can be assumed to be insignificant. 2. No attractive or repulsive forces exist between gas particles. 3. Gas particles are in constant random motion, traveling in independent straight paths until they collide with each other or other objects.

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14.4 Gases: Mixtures and Movements

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