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many protons, neutrons, and electrons are in an atom of cobalt? $P = 27E = 27N = 32$. *Electrons = protons, protons = atomic number, neutrons = atomic mass - protons.

Chemistry Study Guide - South Panola School District

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1) each element is made of atoms. 2) all atoms of a given element are identical there are isotopes. 3) atoms are neither created nor destroyed. 4) atoms are more than 1 element form a compound. Rutherford.

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A. Covalent bonds form when an electron is completely lost or gained from an atom. B. A covalent molecule contains one or more covalent bonds. C. A single covalent bond is drawn as a line between two atoms. D. A pair of electrons is shared between two atoms for each covalent bond.

Chapter 2 Study Guide Basic Chemistry Flashcards | Quizlet

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