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Electrons-are very light particles that orbit around the nucleus at the center of the atom, in layers called energy levels or shells. The electrons are the black particles in the diagram above. Protons - are found inside the nucleus of the atom. The protons are the red particles shown in the diagram. The nucleus is the centre of the atom.

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Inside the Atom

atoms of the same element that have different numbers of neutrons. mass number. the number of neutrons plus protons in the nucleus of an atom. average atomic mass. the mass of the mixture of the isotopes for an element. radioactive decay. the release of nuclear particles and energy. transmutation.

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ANSWER KEY 2 Answer Key 2. cell phones and computers (p. 17) After You Read (p. 18) 1. Students write a sentence explaining how they use information technology. A sample would be: "Searching the Internet for information for a history project is a way of using information technology."

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406 CHAPTER 14 Inside the Atom A

Strange Shadow An electrode is a piece of metal that can conduct electricity. One electrode, called the anode, has a positive charge. The other, called the cathode, has a negative charge. In the tube that Crookes used, the metal cathode was a disk at

How Are Charcoal Celebrations

Inside the Atom Worksheet. 1. Start by selecting the element with 60 protons. 2. Select the Bohr View mode for Neodymium and the Detail View. Look at the tile and find the atomic mass. 3. The atomic mass of an atom is measured in atomic mass units, abbreviated amu. Each proton or neutron weighs one amu, and so the mass of an atom is equal to the total number of protons and neutrons in the nucleus.

Inside the Atom Worksheet - zSpace

The Atom. Worksheet. 1. Particles that are found in the atomic nucleus are: Protons and electrons. Protons and neutrons. Neutrons and electrons. Protons, neutrons and electrons. 2.

Quiz & Worksheet - The Atom | Study.com

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