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# Cold Cook Method Step By Step

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exothermic / endothermic (circle one) reaction and its energy is positive / negative (circle one). b. Draw a line segment that represents the potential energy of the reactants and label the line c. Draw a line segment that represents the activation energy of the forward reaction and label the line. d. Label where the activated complex occurs ...

### HW - Energy Diagram Worksheet

Diagram # potential energy diagram worksheet answers 1 which of the letters a f in the diagram represents the potential draw an energy diagram for a reaction label the axis pe of reactants 350 kJ mol ea 100 kJ mol pe of products 250 kJ mol 7 if8766 potential energy diagram answers chemistry if8766

### Chemistry If8766 Answer Key Potential Energy Diagram

Potential Energy Diagram Worksheet  
ANSWERS 1. Which of the letters a-f in the diagram represents the potential energy of the products?   e   2. Which letter indicates the potential energy of the activated complex?   c   3. Which letter indicates the potential energy of the reactants?   a   4. Which letter indicates the activation

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Chemistry 12 Potential Energy Diagrams  
Worksheet Name: Date: Block: USE THE  
POTENTIAL ENERGY DIAGRAM TO  
ANSWER THE QUESTIONS BELOW: 1. Is  
the overall reaction as shown exothermic  
or endothermic? 2. What is the

activation energy for the forward reaction? 3. What is the activation energy for the reverse reaction? 4.

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contains the following: Converting units  
practice; Foundation level questions;

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Potential Energy Diagrams. The potential  
energy diagram can illustrate the

mechanism for a reaction by showing  
each elementary step of the reaction  
with distinct activation energy (see  
figure below). Figure 18.15. 1: The  
potential energy diagram shows an  
activation energy peak for each of the  
elementary steps of the reaction.

### **18.15: Mechanisms and Potential Energy Diagrams ...**

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- Every chemical reaction will either absorb or release energy.
- A potential energy diagram shows the change in energy during a reaction.
- Heat of Reaction,  $\Delta H$ : the overall difference in potential energy between the products and the reactants.  $\Delta H = \text{PEP} - \text{PER}$