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 redox reactions. An
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 a substance to be
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 electrons, it itself gains
 these electrons and so
 becomes reduced e.g.
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 a substance to be
 reduced, that is to ...

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 Oxidation in terms of
 electron transfer.
 Oxidation is the loss of
 electrons; The
 substance which loses
 electrons is said to be
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 agent: This is the
 substance which
 causes another
 substance to be
 oxidised i.e. it takes
 electrons; Examples of
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 2 O 2) - Used to bleach
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chapters are really
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