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# Finsa Santa Rosala Reseas

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2. Record the number of each atom in each molecule, then record the total number of atoms in the molecule: 1) NaOH 2) HNO<sub>3</sub> 3) Li<sub>2</sub>O  
 Element # of Atoms Element # of Atoms Element # of Atoms  
 Total Total Total 4) Li<sub>2</sub>SO<sub>4</sub> 5) NH<sub>4</sub>Cl 6) CaClO<sub>3</sub>  
 Element # of Atoms Element # of Atoms Element # of Atoms

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Count the number of atoms of each element in the following compounds: CuSO<sub>4</sub> 4 Li<sub>2</sub>CO<sub>3</sub> 3. Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> 4Na<sub>2</sub>CO<sub>3</sub>. Elements Present . Number of Atoms

**Rules For Counting Atoms... - Mr. Hayward's Science Page**  
 Q. How many atoms of aluminum are on each side of the following equation: 4Al + 3O<sub>2</sub> --> 2Al<sub>2</sub>O<sub>3</sub>. answer choices. 2.

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 2KN<sub>3</sub> 9MgSO<sub>4</sub> Sr = Ca = Al = Au = 3D 30 n (Continued)  
 INSTRUCTIONS: Write the quantity of atoms of element opposite the formula of the compound for the quantity of formula units and molecules shown: ) 24 For example: 5PzO<sub>3</sub> For 4Zn(NO<sub>3</sub>)<sub>2</sub> (5 x 2 = ) 10 32. D 20 Mn = —(4x1 x So Inc) 19.

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Multiply with the 5 in front gives us 1 × 5 = 5 Fe. Next, for C, it's slightly more complicated. There are 2 C, but since there's a 3 outside the bracket, it means 2 × 3. And finally, because there's that 5 in front of the entire molecule, total atoms is 2 × 3 × 5 = 30 C.

### Counting Atoms in a Formula - ChemSimplified

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