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and provides the major source of energy for body tissues. When levels are low, we feel hungry.

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1. CHNOPS- most common elements in all living matter. 2. Bonds- ionic (transfer electrons), covalent (sharing- polar/unequal sharing and non-polar/equal sharing), hydrogen (weak bonds between hydrogen and negatively charged items), hydrophobic interactions (how non-polar compounds congregate together- lipids) 3.

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BIOLOGY FINAL STUDY GUIDE CHAPTER 2 THE CHEMISTRY OF LIFE 1. How does the structure of water contribute to its unique properties? 2. What charge is water? 3. What is polar molecule? 4. What are the properties of water? 5. How many protons does water have? 6. How many protons does the oxygen in water have? 7. How many protons does the hydrogen in water have? 8. Why?

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Biology Final What are catabolic reactions? Large molecules are broken down into small ones. Energy is released. Are central to biological processes such as cellular respiration and the digestion of food molecules What are anabolic reactions? Small molecules are assembled into larger ones. Energy is required.