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© UCLES 2005 0620/03/M/J/05 For Examiner's Use (c) The unpleasant smell is caused by carboxylic acids. Bacteria cause the fats on the skin to be hydrolysed to these acids. Silver kills the bacteria and prevents the hydrolysis of the fats. (i) Fats are esters. Give the name and structural formula of an ester. name [1] structural formula [1]

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Step 1: Add a small amount of zinc oxide to some hot dilute nitric acid, and stir. Step 2: Keep adding

zinc oxide until it is in excess. Step 3: Remove the excess zinc oxide to leave colourless zinc nitrate solution. Step 4: Evaporate the zinc nitrate solution until it is saturated.

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© UCLES 2020 0620/21/M/J/20 12 The equation for the complete combustion of methane gas is shown. $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$ Bond energies are shown. bond bond energy in kJ / mol C-H 412 H-O 463 C=O 743 O=O 498 What is the overall energy change, in kJ / mol, for the above reaction? A -1192 B -694 C +694 D +1192

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