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To solve for density, simply follow the equation 1.5.1. For example, if you had a metal cube with mass 7.0 g and volume 5.0 cm³, the density would be $\rho = \frac{7 \text{ g}}{5 \text{ cm}^3} = 1.4 \text{ g/cm}^3$. Sometimes, you have to convert units to get the correct units for density, such as mg to g or in³ to cm³.

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Density Problems Worksheet Middle School - DSoftSchools

Finding Volume from Density and Mass $V = \frac{M}{D}$. $V = \frac{96.5\text{g}}{2.7 \text{ g/cm}^3}$. $V = 35.7 \text{ cm}^3$. Solving Density Problems. If 96.5 grams of gold has a

volume of 5 cm³, what is the density of gold? $D = \frac{M}{V}$ Substitute values into formula Solve $D = \frac{96.5\text{g}}{5 \text{ cm}^3}$. $D = 19.3 \text{ g/cm}^3$.

Solving Density Problems - Slinger, WI

WORD PROBLEMS. 1. A block of aluminum occupies a volume of 16.0 mL and weighs 20.5 g. What is its density? $\frac{20.5 \text{ g}}{16.0 \text{ mL}} = 1.28 \text{ g/mL}$. 2. Mercury metal is poured into a graduated cylinder that holds exactly 32.5 mL. The mercury used to fill the cylinder weighs 406.0 g. From this information, calculate the density of mercury. $\frac{406.0 \text{ g}}{32.5 \text{ mL}} = 12.5 \text{ g/mL}$. 3.

Density Calculations Worksheet I

Answer: To calculate density, use the formula, Density= Mass/ Volume 2. What is the density of water? Answer: The density of water is 1g/ml. 3. How can you determine if an object will float or sink? Answer: If the object has a larger density than that of water (density > 1), it will sink...if it has a density less than 1, it will float. 4.

How To Solve Density Problems? - Density & Buoyancy

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