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2 Chemistry

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Chemical equilibrium, a condition in the course of a reversible
chemical reaction in which no net change in the amounts of

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Given the equation, $\text{N}_2(\text{g}) + 3\text{H}_2(\text{aq}) \rightleftharpoons 2\text{NH}_3(\text{g})$, Find Q and determine which direction the reaction will shift in order to reach the state of chemical equilibrium. Solution: Given, $[\text{N}_2] = 0.04\text{M}$, $[\text{H}_2] = 0.09\text{M}$, and $K = 0.040$

Chemical Equilibrium - Types, Problems, Factors Affecting

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Third, we have chemical equilibrium, equating the chemical species of each comonent in the two phases, $\sum \mu_\alpha = \sum \mu_\alpha$ for $\alpha = 1$ to N_c , (3)

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