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Solutions Worksheet 2 Answers

Solutions to Worksheet #2 1. Step 1: 23lbs / 100gal (original amounts) = .23 (ratio) Step 2: New salt? = 500gals x .23 = 115lbs 2. Step 1: 30lbs (fine) to 60lbs (coarse) Step 2: 30lbs/60lbs = .50 OR 1lb : 2lbs . 1/2 + 1/3 = 3/6 + 2/6 = (3 + 2) / 6 = 5/6 of a drum 4. 3/8lbs - 1/4lbs = 3/8lbs - 2/8 lbs = 1/8lbs, remove 1/8lbs aggregate 5

Solutions to Worksheet #2

Solutions Worksheet #2. (Molarity, Dilutions, Percent Solutions, Molality Problems) Molarity. Tell how you would prepare a 500. mL of 0.50 M ammonium carbonate solution. Include all necessary equipment and amount of chemical (in grams).

Solutions Worksheet #2 - Georgetown ISD

Molarity Worksheet # 2 identifiera ____ What does molarity mean? Number of moles of solute. 1 liter solution. What is the molarity of a solution that contains 4.53 moles of lithium nitrate in 2.85 liters of solution?

Molarity Worksheet 2 ANSWERS - Google Docs

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Worksheets solutions

Dilutions Worksheet – Solutions 1) If I have 340 mL of a 0.5 M NaBr solution, what will the concentration be if I add 560 mL more water to it? 0.19 M (the final volume is 900 mL, set up the equation from that) 2) If I dilute 250 mL of 0.10 M lithium acetate solution to a volume of 750 mL, what will the concentration of this solution be?

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Solutions Worksheet On the line at the left, write the letter of the definition that best matches each term. ____ 1. solution ____ 2. solute ____ 3. solvent ... substance that does the dissolving in a solution Answer each of the following questions in the space provided. 8. Describe the properties of a solution

Solutions Worksheet - Socorro Independent School District

This is the most common type of chemical equation. In this chemical equation, a new product is formed by combining two to three combinations of reactants. For instance, H 2 + O 2 H 2 O. This is a chemical equation where two atoms of hydrogen are combined to form a product, water. This is why this reaction is called as synthesis reaction.

49 Balancing Chemical Equations Worksheets [with Answers]

Box your answer with proper sig digs and units and show work for partial credit. 1. What is the volume in milliliters of an aqueous 0.90 M sodium bromide solution with 5.6 g of solute? 2. What is the concentration of an aqueous sodium bromide solution with 10.7 g of solute dissolved in 1.37 L of

water? 3.

Aqueous Solutions Worksheet - wsfcs.k12.nc.us

Solutions Worksheet. 1) Why does water have such a low vapor pressure? Explain. The hydrogen bonds in water are strong enough that they keep molecules from leaving the surface of the liquid and entering the vapor phase. 2) Give one example of surface tension you’re familiar with, and one example of a surfactant around your house.

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