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Acids are named by changing the ending of the alkane to -oic acid. i.e. a 4 carbon acid goes from butane to butanoic acid Acid salts are named by changing the acid ending to -oate i.e the

4 carbon acid goes from butanoic acid to sodium butanoate
Esters are named by changing the ending of the alkane to -oate and by using alkyl name i.e. the 4 carbon ester below is methyl butanoate(methyl for OCH₃ portion) O O
Anhydrides are dehydrated esters that replace -oic acid with word anhydride Acyl ...

Chapter_17_Study_Guide - 1 Chapter 18 Study Guide 1 ...

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more.... organic bases found in plants that are often poisonous.

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Carboxyl, amine, hydrogen and variable side group Describe two forces that help shape an amino acid chain? Peptide bond holds chains of amino acids together by linking amine and carboxyl groups; hydrogen bonds help fold proteins into shapes and hold them together. Shape depends on amino acids, polarity and R groups.

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- The most common strong acids are HCl, HBr, HI, HNO₃, HClO₃, HClO₄, and H₂SO₄.
- Strong acids are strong electrolytes. •All strong acids ionize completely in solution.

•Example: Nitric acid ionizes completely in aqueous solution. $\text{HNO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{NO}_3^-(\text{aq})$

•Since H^+ and H_3O^+ are used interchangeably, we write HNO_3

AP Chemistry— CHAPTER 16 STUDY

GUIDE Acid-Base Equilibrium

There are 20 amino acids, but only 4 nucleotide bases in DNA. That means it takes at least 3 nucleotides to code for one amino acid, hence a codon (three amino acids). □ The code is read as an RNA code, 5' to 3'.