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SECTION 1 - ^1H NMR

Why compounds absorb radiowaves (background information beyond specifications) NMR (nuclear magnetic resonance) is a very powerful tool for identifying compounds. H, The nucleus of some atoms has nuclear spin (e.g. ^1H , ^{13}C , ^{19}F , ^{31}P), although many atoms do not have any nuclear spin (e.g. ^{12}C).

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Propene reacts with HBr to form H. H reacts with sodium hydroxide to form I, and I reacts with warm acidified potassium dichromate (VI) to form J. The infra-red spectra of H, I and J are given below, but it

does indicate which is - which. Identify the three compounds H, I and J, using the infra-red spectra below, and decide which spectrum belongs

IR TASK 1 - Weebly

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www.CHEMSHEETS.co.uk 17-Jul-12 Chemsheets A2 029 9 IR TASK 2 1) The IR spectra of six compounds are shown. The compounds are: butanoic acid butanone but-3-en-1-ol

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OH-7 Acid = H +, base = HCO 3-8 Acid = H