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Plants protect themselves against self-induced air pollutants

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Weathering - Wikipedia

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Atmospheric chemistry and transport models (CTMs) usually consider two main pathways of particulate nitrate production: (1) gas phase oxidation of NO 2 by OH radicals forming HNO 3, which partitions into the particle phase, and (2) gas phase reactions of NO 2 with O 3 leading to the formation of nitrate radicals and N 2 O 5, followed by hydrolysis of N 2 O 5 in the aqueous phase.

New Multiphase Chemical Processes Influencing Atmospheric ...

Carbonyl groups are found in a variety of organic compounds, and they can be produced by the combustion of fossil fuels and the photochemical oxidation of hydrocarbons in the atmosphere. 37 Carbonyl compounds are of great interest because they are a key source of OH radicals (a major initiator of the oxidation of a variety of pollutants in the atmosphere). 38 Moreover, the hydrolysis of carbonyl species can produce diols, which are highly hydroscopic and have significant implications in ...

Catalytic and autocatalytic chemical processes in the ...

Part 1: Oxidation and OH radicals Oxidation in the Atmosphere Many chemical compounds are emitted into the atmosphere but removal processes prevent them accumulating in the air. Species are removed by dry deposition of gases or particles or can be incorporated into rain and removed by wet deposition. For gas phase organic chemicals,

Part 1: Oxidation and OH radicals - Harvard University

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the atmosphere by contributing to ozone formation^{1,2}, serving as the primary sink for oxidants in the atmosphere^{3,4} and constituting a substantial fraction of global submicrometre particulate matter^{5,6}. Organic carbon (OC) enters the atmosphere primarily as high-volatility gases. Oxidation of these compounds