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dashboard.orthofill.com 9. electron "b" would be less than the IE1 of electron "a". a) Be has 4 protons in its nucleus. b) two c) two d) $4 - 2 = 2$ e) The core charge is equal to the number of valence electrons for a neutral atom. Be has a higher core charge (+2) than Li (+1) and a higher IE1 than Li.

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55Exercises 1. a) Based on our revised shell model, how many peaks would be expected in a photoelectron spectrum of lithium? b) What would you expect the relative sizes of the peaks to be? c) The 1s ionization energies for H, He, and Li are 1.31, 2.37, and 6.26 MJ/mole, respectively.

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Answers to ChemActivity 1 – The Nuclear Atom- Chapter 5 1. 6 2. 6,7,7 3. 6,6,7 4. (a) neutral same e and p, ion different e and p (b) assign +1 to each proton and -1 to each electron and take the difference. If more -1 it is a negative ion, if more +1 it is a positive ion. 5.

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