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Oxidation-reduction reactions are of central importance in organic chemistry and biochemistry. The burning of fuels that provides the energy to maintain our civilization and the metabolism of foods that furnish the energy that keeps us alive both involve redox reactions. Figure \(\backslash\PageIndex{1}\): The

Burning of Natural Gas.

13.2: Redox Reactions in Organic Chemistry - Chemistry ...

Autoxidation (sometimes auto-oxidation) refers to oxidations brought about by oxygen at normal temperatures, without the intervention of flame or electric spark. The term is usually used to describe the degradation of organic compounds in air. Many common phenomena can be attributed to autoxidation, such as

food going rancid, the 'drying' of varnishes and paints and the perishing of rubber.

Autoxidation - Wikipedia

Antioxidants are particularly important in the context of organic chemistry and biology. All living organisms maintain a reducing environment inside their cells, all cells contain complex systems...