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~~Saturated,~~
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The solubility of a solid

in water increases with

an increase in

temperature.

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Solubility Temperature
Graphs Chapter 14
Answers Use the
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For questions 1 – 4 an
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Graphs Chapter 14
Answers Make a line
graph to plot the data
from the table.

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Answers Use the
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Answers solubility is
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Na₂SO₄'s solubility
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about 14 g, meaning
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recrystallize. Summary
The solubility of a solid
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an increase in
temperature. Chapter
14: Solutions - Oneonta

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Answers relationship
between solubility (in
grams of solid per
volume of water) vs
temperature. If the
solution is above the
solubility line it is
supersaturate and

below the solubility line
it is unsaturated. Points
along the line are
points of saturation.
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solubility temperature graphs chemistry matter Solubility Versus Temperature This chart shows the solubility of various substances in water at a variety of temperatures (in degrees Celsius). Notice how NaCl's

solubility is relatively constant regardless of temperature, whereas Na₂SO₄'s solubility increases exponentially over 0–35 degrees Celsius and then abruptly begins to decrease. Solid Solubility and Temperature | Introduction to Chemistry

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Celsius and then abruptly begins to decrease. Solubility Temperature Graphs Chapter 14 Worksheet Answers Solubility Temperature Graphs Chapter 14 Answers Use the Page 1/4

Solubility Temperature Graphs Chapter 14

Worksheet Answers
on the solubility of NaCl. 8. Explain how you might make a solution containing 42 g KCl dissolved in 100 g H₂O at a temperature of 40°C. What term describes this type of solution? Solubility-Temperature Graphs TEACHING TRANSPARENCY WORKSHEET Use with Chapter 14, Section 14.3 42 Substance Solubility at 10°C Calcium chloride (CaCl₂)

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Worksheet Answers
1. Which solid has the lowest solubility at 60 °C? 2. Which solid has the same solubility as potassium nitrate at 52 °C? 3. Which solid has a solubility that changes least with

temperature? 4. Which solid would give a deposit of 20 g if a saturated solution in 100 g of water at 60 °C was cooled to 20 °C? 5.

Solubility Curves
(solutions,
examples, activities

...

Table

\\(\\PageIndex{1}\\):

Solubility of table salt

as a function of water. In this graph 6 measurements of the solubility were made at different temperatures. Here, the solubility is the maximum amount of salt that can be dissolved in 100 grams of water, and any extra salt beyond that falls to the bottom of the container as a precipitate (solid).