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1. True: b), c) and d) 2. Reaction = 2 x
 reverse of original reaction. So $K_2 = (1/K_1)^2 = (1/0.447)^2 = 5.00$. 3. $K = [NH_3]^2 / [N_2] \times [H_2]^3$; more product
 made. Increasing the pressure increases
 the concentration; counteract this by
 shifting to the side with fewer moles.

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 the space at the left, write the letter of
 the word or phrase that best completes
 the statement or answers the question.

1. The freezing point of water is an example of a.

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13.1. Classify the following amines as primary, secondary and tertiary: Ans: (i) 1° (ii) -3° (iii) 1° (iv) 2°. 13.2. Write the structures of different isomeric amines corresponding to the molecular formula, C₄H₁₁N. (i) Write the IUPAC names of all the isomers.

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