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- Expresses the relationship between the pressure, volume and temperature of a fixed amount of gas.
- $PV/T = K$ or $P_1V_1/T_1 = P_2V_2/T_2$ Ex: A sample of gas has a volume of 201 L when its temperature is 293 and its pressure is 24 mm Hg. What volume will the gas occupy at STP?

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Think about these steps to answer the question: Identify the system - P.E.). goes from liquid PE change, In this This is an type of change, the system (the water) heat from the surroundings. Identify the surroundings - This reaction is When Act E is high, the reaction is (slow, fast). Sketch a diagram of these reactions: thermic. faster, exothermic

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