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CH 3 OH = 32.05 grams/mole H 2 O = 18.02 grams/mole moles alcohol = 100 g/32.05 g/mole = 3.12 mole moles water = 100 g/18.02 g/mole = 5.55 mole. 26. The amount of CuCl is 0.00152 mole. If the powder dissolved completely, the solution would be 0.00152 molar with respect to both Cu + and Cl -.

Answers to Chemistry Problems - CliffsNotes

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As in any equation, the elements on the left side of the equation (the reactants) must equal the elements on the right (the products). Here are just a few things to remember when balancing chemical equations: Same elements in equal numbers: For an equation to be balanced, it must have the same number of each element on each side of the equation. The mass of the reactants has to equal the mass of the products — remember the law of conservation of mass?

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Chemistry Answers - Assignment Expert

Chemistry Solutions Practice Problems 1. Molar solutions. a. Describe how you would prepare 1 L of a 1 M solution of sodium chloride. The gram formula weight of sodium chloride is 58.44 g/mol. Answer: To make a 1 M solution of sodium chloride, dissolve 58.44 g sodium chloride in 500 mL water in a 1000-mL volumetric flask. When all the solid is dissolved and the solution is at room temperature, dilute to the mark and invert the flask several times to mix.

Chemistry Solutions Practice Problems | Carolina.com

Chemistry is the science of matter: its composition, its properties, the changes that lead to its formation, and the ways it interacts with other matter in its surroundings. ... PFG Buffer Recipes - This simple tool delivers fast and accurate answers, ... Plug in the equations values in the text boxes and your problems are solved.

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Problem #1: FeCl 3 + MgO ---> Fe 2 O 3 + MgCl 2. Solution: 1) Balance the Cl (note that 2 x 3 = 3 x 2): 2FeCl 3 + MgO ---> Fe 2 O 3 + 3MgCl 2. The Fe also gets balanced in this step. 2) Pick either the O or the Mg to balance next: 2FeCl 3 + 3MgO ---> Fe 2 O 3 + 3MgCl 2. The other element (Mg or O, depending on which one you picked) also gets balanced in this step.

ChemTeam: Balancing Chemical Equations: Problems #1 - 10

This category encompasses all questions about the composition, structure, properties, and reactions of matter.

Answers about Chemistry

In either case, the answers are on the web page. * Web page: Naming chemicals. Includes an expanded introduction and answers with explanations. * pdf file: Naming chemicals handout. The original class handout, with a brief introduction and the questions. Use the web site, just above, for answers and explanations. Mass-mole interconversions.

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Answers To Mastering Chemistry Problems

Answers . 1. 1.587 grams per milliliter 2. 407.5 grams 3. 559 milliliter 4. 1051.2 grams 5. 26561 grams or 26.56 kilograms 6. a. 7.8 grams per milliliter b. 3 x 10-2 kilograms/centimeter 3 7. Liquid A. (0.75 grams per milliliter) 8. 68 kilograms 9. 8.33 pounds (2.2 kilograms = 1 pound, 1 liter = 0.264 gallons) 10. 483.6 centimeters³