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Chapter 20: Coordination Chemistry: Reactions of Complexes 131

Chapter 20 Coordination chemistry: reactions of complexes Redox reactions Ligand substitution in octahedral complexes Ligand substitution in square-planar complexes Ligand substitution reactions Photochemical reactions. 12-7 The trans Effect 12-8 Oxidation-Reduction Reactions 12-4 Experimental Evidence in Octahedral substitution 12-5 Stereochemistry of Reactions 12-6 Substitution Reactions of ...

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coordinationThe reaction of one or more ligands with a metal ion to form a coordination compound. redoxA reversible chemical reaction in which one reaction is an oxidation and the reverse is a reduction. donor atomThe atom within a ligand that is bonded to the central atom or ion within a coordination complex.

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Coordination complexes can undergo a variety of reactions, including electron transfer, ligand exchange, and associative processes. Key Terms. coordination: The reaction of one or more ligands with a metal ion to form a coordination compound. redox: A reversible chemical reaction in which one reaction is an oxidation and the reverse is a reduction.

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The first order reaction appears to be a dissociative reaction or a solvent- assisted dissociation of CO, followed by a fast addition of $\text{As}(\text{C}_6\text{H}_5)_3$. The other path shows first

CHAPTER 12: COORDINATION CHEMISTRY IV: REACTIONS AND ...

Coordination chemistry: reactions of complexes. Reactions of Complexes. Typically measure ligand substitution reactions in solution (usually water) Lability and Inertness Labile: complexes with half -lives under 1 minute . Inert: complexes with half-lives longer than 1 minute (better term is non-

labile) Figure 20.1 shows lifetimes for exchange of water

Chapter 21 Coordination chemistry: reactions of complexes

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Chapter 5. Thermochemistry – Chemistry

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